

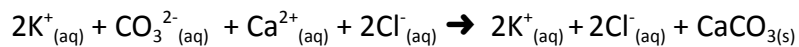
Chemistry 20: Non, Total, Net ionic reactions

1. Numerical response question

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Left justify your answer in the boxes provided

Consider the following total ionic equation.



Identify the spectator ion(s) in this reaction from the list below:

1. $\text{K}^+_{(\text{aq})}$
2. $\text{CO}_3^{2-}_{(\text{aq})}$
3. $\text{Ca}^{2+}_{(\text{aq})}$
4. $\text{Cl}^-_{(\text{aq})}$
5. $\text{CaCO}_{3(\text{s})}$

Solutions:

1. 14