Chemistry 20: Balancing Reactions

1.	Numerical response question
	Left justify your answer in the boxes provided.
	Determine the smallest whole number coefficients that will balance the following reaction.
	$C_3H_{8(g)} + CO_{2(g)} \rightarrow CO_{2(g)} + H_2O_{(g)}$
2.	Numerical response question
	Left justify your answer in the boxes provided
	Consider the double replacement reaction below. NaBr _(s) + Al(OH) _{3(s)} \rightarrow AlBr _{3(s)} + NaOH _(s)
	The smallest whole number coefficients to balance this reaction will be,,, and
3.	Numerical response question
	Left justify your answer in the boxes provided
	Find the <u>SUM</u> of the smallest whole number coefficients in the balanced equation of the complete combustion of propane $(C_3H_{8(g)})$

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Solutions:

- 1. 1534
- 2. 3113
- 3. 13